Ce-Cr/硅藻土对几种低分子量有机物 氧化作用的催化活性研究*

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摘娶 考察了添加 Al₂O,对硅藻土的孔容、平均孔径及比表面的影响。并研究了负载型 Ce-Cr、Ce-Cr-Mn 等复合氧化物催化剂对丙酮、正丁胺、甲苯氧化反应的活性。氧化活性的顺序为: Ce-Cr-Mn-K≈Ce-Cr>Mn>Cr>Ce. 而且能够抑制正丁胺氧化过程 NO。的形成,正丁胺氧化 NO。控制能力为: Ce-Cr-Mn-K>Ce-Cr-Mn>Ce-Cr>Ce.

关键词 硅藻土,催化剂,氧化活性.

我国硅藻土矿藏资源极为丰富,据报道,1985年探明储量在 4 亿吨以上,仅浙江省就达6000万吨,原土年产量5000吨,基于这个情况,当前大力推广硅藻土在各个领域的应用具有十分重要的意义。硅藻土主要成分为无定形SiO₂,并含有相当数量的石英、蒙脱石、高岭土和少量有机质等。由于天然硅藻土具有多孔性结构以及耐热、耐酸等特点,所以是催化剂常用的一种载体。但因机械强度较差,目前主要用作固定床催化剂载体。在燃烧催化剂载体方面的应用,国内还未见报道。本文用添 Al₂O₃ 的硅藻土为载体,考察了负载 Ce、Cr、Mn 等复合氧化物对丙酮、甲苯、正丁胺的氧化反应性能,其目的在于开拓硅藻土的应用范围。

1 实验部分

1.1 原料

浙江嵊县硅藻土(水洗土);铝溶胶,浙江温州催化剂厂产品;其它试剂: Mn(NO₃)₂ (50%液体)、 Cr(NO₃)₃•9H₂O、 Ce(NO₃)₃•6H₂O、KNO₃ 均为化学纯.

1.2 硅藻十成型

硅藻土用于燃烧催化剂载体必须重新成型. 当硅藻土预先经 H₂SO₄ 处理后,尽管加入大量铝溶胶(达 30% Al₂O₃)粘合,发现强度很

差,极易粉碎。因此本研究直接用水洗硅藻土, 添加铝溶胶粘接成型,120℃烘干,800℃焙烧 3h,破碎成 3—5mm。

1.3 催化剂制备

根据配制量称取一定量的活性组分 物 质,加适量水溶解并调整到硅藻土载体的吸 水 值,称取规定量的上述载体进行浸渍,强迫吸附,最后于 500℃ 空气氛中活化 2h。催化剂以活性组分元素符号表示。

1.4 催化剂活性评价

评价装置及分析方法见文献[1],反应管内径 $20 \,\mathrm{mm}$,催化剂装量 $30 \,\mathrm{ml}$, 空速 $10000 \,\mathrm{h^{-1}}$,反应物浓度:丙酮 $6-7 \,\mathrm{g/m^3}$;正丁胺 $2-3 \,\mathrm{g/m^3}$;甲苯 $4-5 \,\mathrm{g/m^3}$ 。NO、NO₂ 用盐酸萘乙二胺比色法测定^[2]。

1.5 主要仪器

Carlo Erba 2000 型压汞仪; Carla Erba 自动吸附仪; QCY-602 型电动强度测定仪.

2 结果和讨论

2.1 添铝量对硅藻土物性的影响

由于活性组分直接依附于载体,载体的表面物理性质在很大程度上决定了催化剂的物理

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性质, 尤其是比表面、孔径、强度。分别添加 0%、3%、6%、12% Al₂O₃ 的硅藻土成型, 800℃ 焙烧后试样的压汞曲线如图 1 所示。从图 1 孔 半径与累积孔容的关系可求得样品的比表面和 平均孔径[3]。 结合其它实验数据列于表 1。由 表 1 可见,4 个试样的强度、BET比表面随ALO, - 的增加而增加;而平均孔径、总孔容及压汞比表 面却随 Al₂O₂增加而下降。样品的 X-射线衍 射分析表明, 随 Al₂O, 增加, Al₂O, 衍射线的 相对强度没有可觉察的变化,这表示 Al₂O₃ 大 多以无定形的 Al₂O₃ 存在。样品的压汞曲线可 见,随 Al₂O, 量增加,小孔径 (37—100 Å) — 边曲线上跷势头明显增加,同时 BET 比表面 明显增加,这意味着小孔随 Al₂O₃ 量的增加而 增加,小孔可能来自 Al₂O₃ 和 Al₂O₃ 进入硅藻 土大孔使大孔变小的那部分。因此可通过 Al₂O₃ 的添量来调节硅藻土载体的物理性质, 以满足 实用要求,

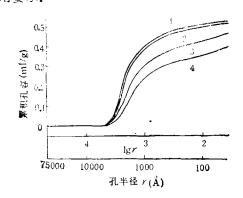


图1 压汞曲线

1.硅藻土 2.3% A₂O₃ 硅藻土 3.6%Al₂O₃ 硅藻 土 4.12%Al₂O₃ 硅藻土

表 1 试样的基本物性

Al.O. 法是 (%)	焙烧 温度 (℃)		压汞法 75000 比表面 (m²/g)	(Å)	BET 比表面 (m²/g)	吸水率 (ml/ 100g)	堆密度 (g/ ml)	压碎 强度 (kg /颗)
0	ļ	0.52	21.3	486	20	60	0.53	9*
3		0.51	20.9	410	42	59	0.55	12
. 6	800	0.46	20.5	403	60	54	0.60	15
12	!	0.39	20.2	389	70	51	0.62	10
				}				-

^{*} 易粉遊

2.2 单组分催化剂的氧化活性

Ce、Cr、Mn 单组分催化剂(载体为 6% Al,O、硅藻土)对丙酮、正丁胺、甲苯的氧化活性见表 2. 反应活性以反应物转化率 >98% 的下限反应温度 T_{98} 表示, T_{98} 越低表示活性越高。

表 2 单组分催化剂的氧化活性*

	T,8(°C)			
催化剂	丙酮	正丁胺	甲苯	
Се	240	340	360	
Cr	300	320	340	
Mn	280	280	320	

^{*} 活性组分负载量 0.08mol/100g 载体。

由表 2 可见,催化剂上有机物氧化活性次序为: 丙酮>正丁胺>甲苯。催化剂对有机物的氧化活性总体上为: Mn > Cr > Ce。

2.3 活性组分负载量及 Al₂O, 添加量对催化 剂活性的影响

对含氮有机物氧化催化剂的研究^山表明,Ce-Cr复合氧化物催化剂,Ce、Cr在一定比例时有明显的协同催化作用,并表现出较高的氧化活性.本文选择 Ce、Cr为活性物质,其摩尔比Ce:Cr = 3:1,考察了 Ce-Cr负载量及硅藻土的 Al₂O₃ 添加量对氧化活性的影响,结果见表3。由表 3 可见,催化剂氧化活性的次序为:

$$B = C = D > E > A$$

袋 3 Ce-Cr 催化剂的氯化活性

		- ,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,				
	催化剂		T 98(°C)			
编号	Ce、Cr 总负载量 (mol/ 100g)	□藻土添 Al ₂ O ₃ 量(%)	树间	正丁胺	甲苯	
A	0.04	6	220	280	350	
В	0.08	6	180	220	300	
С	0.12	6	180	220	300	
D	0.08	3	180	220	300	
Е	0.08	12	200	240	300	

Ce、Cr 的负载量从 0.04mol/100g 增高到 0.08 mol/100g, 催化剂活性明显提高, 但再增加并无益处。 同一负载量 (0.08mol/100g) 载于不

同含 Al₂O₃ 量的硅藻土上,催化剂的活性总体上 Al₂O₃ 含量低时较佳。但从机械强度等物理性质考虑,Al₂O₃ 添加量 6% 时为最佳,故本文以下均采用 Al₂O₃ 添加量 6% 的硅藻土载体。 2.4 添加 Mn、K 对 Ce-Cr 催化剂活性的影响

表 4 Mn、K 对 Ce-Cr 催化剂氧化活性的影响

	T ₉₈ (°C)			
丙酮	正丁胺	甲苯		
180	220	300		
200	220	300		
180	220	280		
200	220	280		
180	220	260		
180	230	280		
180	240	300		
	180 200 180 200 180 180	内嗣 正丁族 180 220 200 220 180 220 200 220 180 220 180 220 180 231		

^{*} Ce: Cr = 3:1 Ce、Cr、Mn 总负载量 0.08/100g K₂O 量按 (Ce、Cr、Mn) 总负载量的百分比计, K_{2.5} 表示 K₂O 添加 2.5%.

鉴于氧化锰是重要的氧化还原型催化剂,是过渡金属复合氧化物催化剂的重要组成部分。氧化钾是强碱性物质,掺钾后使催化剂表面酸碱度发生变化,从而改变催化性能。表4是 Ce-Cr 催化剂添加 Mn、K 后的氧化性能。由表4可见,添 Mn 没有改变 Ce-Cr 催化剂的氧化活性,但在此基础上再添 K,随 K 量增加丙酮活性不变,正丁胺活性呈下降趋势,而少量 K 对甲苯氧化有促进作用。

2.5 控制 NO, 生成能力

对含氮有机物(如正丁胺)的氧化、氨基的氧化有两种可能,即变成 N₂或 NO₂,作为净化处理,保护环境的要求,希望有控制地氧化到 N₂,避免 NO₂ 的生成,以免造成二次污染.因此,正丁胺氧化过程 NO₂ 控制率是一个重要指标. Ce、Ce₂Cr₁、Ce₅Cr₁,Mn₃₀、Ce₅、Cr₁,Mn₃₀K_{2.5} 上正丁胺完全氧化时 NO₂ 的控制能力见表 5。可见对于含氮有机物,要求催化剂一方面有较高的氧化活性(T₉₈ 低),另一方面具有高的控制 NO₂ 生成能力。即氧化活性温度(T₉₈)与 NO₂

明显生成温度(NO, 控制率 >90% 的上限温度 T_{90NO} 、表示)之间要有较大的温度范围(ΔT),我们把 ΔT 的大小用来衡量催化剂 NO, 控制能力, ΔT 越大表示 NO, 控制能力越强。由表 5 可见,在 Ce-Cr 上添加 Mn、K 尽管对正 丁胺氧化活性没有改善,但 NO, 控制能力明显提高,其中以 Ce-Cr-Mn-K 4 组分的性能最佳。

表 5 NO, 控制能力

				~ ====	
催化剂	温度(℃)	NO,控 制率 (%)*	T 98(°C)	T 90 NO _x	T ONO TO 8
Ce	360 380	90.2 78.1	340	360	20
Ce ₃ Cr ₁	260 280 350	93.2 83.2 56.2	220	260	40
Ce,3Cr ₁₇ Mn ₃₀	290 300 350	92.5 90.1 68.7	220	300	80
Ce ₅₃ Cr ₁ , - Mn ₃₀ - K _{2.5}	320 340 360	92.5 90.3 78.9	220	340	120

^{*} 产物中 NO, 以 NO, 计

NO, 控制率=NO, 理论生成量-实测量 ×100% NO, 理论生成量

上述结果可见, Ce-Cr-Mn 复合氧化物催化剂对丙酮、正丁胺、甲苯有较高的氧化活性. 具有工业应用前景。当然本文仅是初步的探索, 对于如何进一步提高催化剂性能及氧化机理的理论问题,还有待于进一步研究,相信硅藻土能在催化燃烧应用领域发挥更大的优势。

3 结论

- (1) 硅藻土的强度及比表面与 Al₂O₃ 添加 量成正比,而孔容、平均孔径与Al₂O₃ 量成反比.
- (2) 添加 Mn、K 对 Ce-Cr 复合氧化物的氧化活性影响不大,但对正丁胺氧化过程 NO。控制能力有明显提高。

参 考 文 献

- 1 袁贤鑫等。石油化工。1990,12: 828
- 2 国家大气环境质量标准 GB3095-82
- 3 王祖兴等. 非金属矿. 1989,6: 9

HUAN JING KEXUE Vol. 14 No. 1, 1993

Abstracts

Chinese Journal of Environmental Science

Study on Water Quality Control Plan of Yongjiang River of Zhejiang Province. Huang Mingrong, He Qiang, Cheng Shengtong, Jing Wenyong et al. (Institute of Environmental Engineering, Tsinghua University, Beijing 100084): chin. J Environ. Sci., 14(1), 1993, pp. 2—6

Yongjiang River flows through Ningbo city, whose flourishing economy makes the river polluted worse and worse. The old industry mixes with the new, and the drainage system in the old district is imperfect. These are the common problems for the coastal cities opening to the outside world, based on the combination of the principles of concentration control for degradable tants, and determination of water quality index system according to its real conditions and conprehensive analysis of pollution control scenari, technologies, reception capacity of pollutants of the environment and economic response of the old and new enterprices, a chart of pollution load-capital cost-environmental effect was dratted. It was concluded that Ningbo needs only to invest 190 million Yuan (58% less than the secondary treatment scheme) and remove BOD load 2.5 t/d from leading sources to keep the water quality for Yongjiang River to its 1987 level.

Key words: Yongjiang-river, pollution control, Programme.

The Effect of COD/SO₂² Ratio on the Amaerobic Digestion of Sulfate Wastewater. Ding Qiong et al. (Department of Environmental Engineering, Tsinghua University, Beijing 100084): Chin. J. Environ. Sci., 14(1), 1993, pp. 7-12

The effect of COD/SO₄² ratio on anaerobic digestion of sulfate containing high strength organic wastewater was studied with batch test method. Experimental results show that COD/SO₄² ratio is an important factor which influences the anaerobic digestion of the wastewater. In this study, when COD/SO₄² was more than 15, the reduction of sulfates hadlittle or basically no inhibition effect on the anaerobic digestion. When COD/SO₄² was between 5 to 15, the reducti-

on of sulfates had slight inhibition effect on the anaerobic digestion and the relative methane-producing rate was 79.2% to 94.7%. When COD/SO₂² was 0.5 to 5, the inhibition effect of sulfate reduction on anaerobic digestion was moderate and the relative methone-producing rate was 61.6% to 79.2% When COD/SO₂² was below 0.5, the anaerobic digestion would be severely inhibited. Experimental results also show that when COD/SO₂² was greater than 1, a limear relationship between relative methane-producing rate and COD/SO₂² ratio existed.

Key words: sulfate reduction, COD/SO₄² ratio, anaerobic inhibitive, relative methane-producing rate.

Fractionation of TOC in Water from Yinluan Channel. Shu Tao, Deng Baoshan et al. (Dept. of Urban & Environ. Sci., Peking University, Beijing 100871): Chin. J. Environ. Sci., 14(1), 1993, pp.13—16

Organic compounds in water from Yinluan Channel was fractiona ted into five fractions of solids, hydrophobic compounds, humic substances, anions, and other dissolved hydrophillic compounds by filtration, adsorption on XAD-8 resin at different pH values, and adsorption on anion exchange resin. The total amount of DOC was found to be 4.75 mgC/L, half of which is humic substances. The fractions were characterized by UV spectrometry and over 90% of the UV absorption is attributed to the fraction of humic substances.

Key words: TOC, DOC, fractionation, humic substances; UV spectrometry.

Study on the Catalytic Activity of Ce-Cr/ Diatomite Catalysts for the Oxidation of Several Organic Compounds. Luo Mengfei, Yuan Xianxin et al. (Dep. of Chemistry, Hangzhou University, Hangzhou 310028); Chin. J. Environ. Sci., 14(1) 1993, pp. 17—19

The influence of addition of alumina on the pore volume, mean pore size and surface area of diatomite was investigated. The activity of Ce-Cr. Ce-Cr-Mn, etc. complex oxide supported catalysts

HUAN JING KEXUE Vol. 14 No. 1, 1993

Abstracts

Chinese Journal of Environmental Science

for the oxidation of acetone, butylamine and toluene was studied. The order of oxidation activity
was determined to be: Ce-Cr-Mn-K ~Ce-Cr-Mn ~
Ce-Cr>Mn>Cr> Ce. The formation of NO_x
in the oxidative reaction of butylamine can be
retarded. The order of NO_x formation retarding ability of the catalysts was: Ce-Cr-Mn-K
> Ce-Cr-Mn> Ce-Cr> Ce.

Key words: diatomite catalyst, oxidation, alumina.

The Effect of Simutating Acid Rain on the Release of Active Aluminum from Soil. Fu Liusong, Wu Jiemin et al. (Dep of Environ. Protection, Zhejiang Agri-University, Hangzhou 310029): Chin. J. Environ. Sci., 14(1), 1993, pp. 20—24

The effect of simulating acid rain on the release of active aluminum from a typical acidic soil of Zhejiang Province was investigated by dymanic leaching experiments under laboratory conditions. Results indicate that much of the active Al would be released from soil if the soil is leached by acid rain with pH value lower than 4.3 over a long period of time: the critical acid rainfall, which leads the guick release of active Al from soil, is inversely proportional to the acidity of the acid rain. Based on the relationship between the critical acid rainfall and the acidity of the acid rain, the possible effect of natural acid rain on the release of active Al from soils can be assessed.

Key words: simulating acid rath, active Al, Soil.

Study on the Analysis of Aldehydes in Ambient Air. Li Xiaozhen, Bai Yuhua et al. (Dept. of Technical Physics, Peking University, Beijing 100871): Chin. J. Environ. Sci., 14(1), 1993, pp.25—28

Two spectrophotometric methods (MBTH method and Acetyl Acetone method) for the determination of formaldehyde (HCHO) and total aldehyde (TCHO) have been studied and optimized for the analysis of the ambient air in Beijing. Results show that time and temperature of color reaction are the major factors which influence obviously the completeness of the reaction. To get the best results, experimental conditions if the reaction

time was changed from 30 min to 80 min at room temperature for MBTH method and from 120 min at room temperature to 5 min at 70°C for Acetyl Acetone method. Other aldehyde is not detected by Acetyl Acetone method, which is natually specific for Formaldehyde. Hovewer MBTH method can be used for the determination of formaldehyde and other aldehyde as well, and the results were expressed in the concentration of formaldehyde equivalents.

Key words: spectrophotometry, formaldehyde, aldehyde.

Granulation Technology of BOF Sloughy Dust in Wet Rotary Kiln. He Tianxiang (Design and Research Institute of Maanshan Iron and Steel Company, Maanshan 243000): Chin. J Environ. Sci., 14(1), 1993, pp.29—31

Preliminary dehydration, granulation and sintering of vacuum-filter dust containing 40% of water can continuously be accomplished in wet rotary kiln. The product from the kiln is a kind of sintered briquette. This sintered briquette can be reused in the steelmaking process as slag-forming material. The fuel consumption of the BOF dust recycling process is not high and economically acceptable.

Key words: BOF sloughy dust, granulation, sintering.

Study on Electrolytic Process for the Treatment of Wastewater Containing Copper Nitrate. Li Chunhua (Beijing Polytechnic University, Beijing 100022): Chin. J. Environ. Sci., 14(1), 1993, pp.32—36

This paper deals with the study on conditions and principles of the recovery of copper from copper-nitrate wastewater by electrolysis. The change of NO_3^- before and after electrolytic process and the approach of raising the limit of current density were also discussed in the paper. Results show that, when the concentration of $NO_3^- \le 1.0$ mol/L, the corresponding Cu^{2+} is in the range of 3-8g/L, H^+ is in the range of 1.0-2.0 A/dm²,