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城市污水再生处理中微量有机污染物控制的关键难题与解决思路 王文龙,吴乾元,杜烨,黄南,陆韻,魏东斌,胡洪营







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污泥 EPS 作为阻燃剂的机制归纳与潜力分析

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摘要:高附加经济价值回收产物胞外聚合物(EPS)提取可以有效推动剩余污泥资源化目标. EPS 除可替代传统藻酸盐应用于食品、医药、纺织、印染、造纸和日化等行业外,它在阻燃方面独特的性能已展现出用作高端航空器防火涂层的诱人潜力. 这得益于 EPS 本身复杂的化学结构与物质成分,其优异的相容性、黏附性和生物合成性等优势使它的阻燃与环保性能可能优于市场目前已应用上各种阻燃剂. 因此,系统分析和总结 EPS 的阻燃特性机制对其广泛应用极具现实意义. 首先,在总结市售现有各类阻燃剂之优缺点基础上,对比分析 EPS 阻燃特性以及应用潜力. 其次,在阐述 EPS 中磷(P)元素与类藻酸盐(ALE)物质阻燃原理的情况下,厘清 EPS 胞外蛋白的协同阻燃过程,以揭示 EPS 阻燃性能的可能机制. 以此为基础上,随之综合评价 EPS 阻燃特性,并与其他阻燃剂横向对比,总结 EPS 作为阻燃剂的优势所在. 最后,为进一步提高 EPS 用作表面涂层阻燃材料性能,提出增加 EPS 磷元素含量、提高和纯化 EPS 中 ALE 以及优化 EPS 与基质的修饰策略.

关键词: 胞外聚合物(EPS); 防火材料; 阻燃特性; 磷(P); 类藻酸盐(ALE); 胞外蛋白中图分类号: X705 文献标识码: A 文章编号: 0250-3301(2021)06-2583-12 **DOI**: 10.13227/j. hjkx. 202010178

Mechanisms Summary and Potential Analysis of EPS as a Flame Retardant

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Abstract: High value-added extracellular polymer substance (EPS) extracted from excess sludge can effectively promote resource recovery from wastewater. EPS can replace traditional alginate in the food, medicine, textile, printing and dyeing, papermaking, and household chemicals industries. Moreover, its unique performance as a flame retardant has shown attractive potential for aircraft including space shuttles. This is due to the complicated chemical structure and composition of EPS, the excellent compatibility, adhesion, and other advantages of which could yield environmental-friendly flame-retardants. Therefore, a systematic analysis and summary on the mechanisms of EPS as flame retardants is of significance for future application. On the basis of the advantages and disadvantages of other fire-resistant materials on the market, the characteristics and application potential of EPS are analyzed and summarized. Second, the possible fire-resistant mechanisms of phosphorus and alginate-like substance (ALE) in EPS are revealed, and the synergistic flame-retardant effects of extracellular-proteins are also elucidated. Based on this, the flame-retardant characteristics of EPS are comprehensively evaluated and compared with other fire-resistant materials. To further improving the performance of EPS as a flame-retardant material, some modification strategies are proposed, such as increasing their phosphorus content, purifying and enhancing the content of ALE in EPS, and optimizing the modification methods of EPS on their substrates.

Key words: extracellular polymeric substance (EPS); fire-resistant material; flame-retardant properties; phosphor (P); alginate-like substance (ALE); extracellular-protein

从视剩余污泥为一无是处的"废物"到认识到 其实污泥是资源与能源的载体的观念转变,使污水 处理走上了可持续发展的目标. 剩余污泥主要由细 胞及其胞外聚合物(EPS)组成,EPS 约占污泥干重 的 10%~40%^[1]. EPS 包含多糖、蛋白质(结构蛋白 或胞外酶)、核酸、脂质、腐殖质及其他部分胞内物 质^[2];这些物质通过静电作用力、氢键结合、离子 吸引力和生物化学等作用形成紧致高密网状结构; 作为微生物的保护层,这些物质可抵御外部重金属 和有毒化合物等不利因素侵袭^[3]. EPS 这种复杂成 分与结构可使其作为高附加值产品回收并应用成为 可能^[4].

有研究表明,提取自好氧颗粒污泥的 EPS 具有较高的经济价值,可用作各类防水和防火材料^[5,6].

实验分析显示, EPS 防火性能符合美国联邦航空条例(FAR)飞机内部阻燃材料阻燃要求. 虽然普通活性污泥 EPS 性能略微逊色,但其亦可媲美甚至替代市场现有大部分阻燃剂^[5],所以研究 EPS 阻燃特性,分析其可能来源于其中所包含的磷元素和有机物(多糖和蛋白质类等)及其复杂的化学结构. 然而,现有研究缺少对 EPS 物质阻燃机制清晰的认识,对其阻燃效果及其未来可能应用方式也未十分明确,这势必会阻碍 EPS 作为高附加值物质的提取、回收及应用. 为此,有必要根据现有研究总结并分析 EPS 作为表面涂层防火材料的作用机制与实

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在总结当前不同阻燃剂应用与实际效果以及发展趋势的基础上,首先阐明 EPS 物质作为阻燃剂应用之前景;其次,深入剖析 EPS 成分与结构特性并总结 EPS 阻燃的机制与过程;随后,与传统阻燃剂比较后归纳 EPS 用作阻燃剂之优势与性能;最后,挖掘可能提升 EPS 阻燃性能的手段与方法.以此为目的,以期为未来广泛深入研发 EPS 防火材料奠定理论与应用基础.

1 阻燃剂应用与效果

各种不同阻燃剂已被广泛应用于塑料与纺织工 业中.目前,市场上使用的阻燃剂多是无机阻燃剂, 主要为氢氧化铝/镁,市场份额达38%;其次是溴化 和氯化物组成的卤化阻燃剂,占比31%;而有机磷 阻燃剂份额仅为18%;其他无机磷、氮和锌基阻燃 剂则更少[7]. 无机氢氧化铝/镁阻燃剂等具有良好 抑烟甚至消烟性能,具有无卤、无毒和不挥发等安 全和环保优点,但它们的应用填充量较大、在加工 成型时流动性差[8],也会降低合成材料的再加工和 机械性能[9]. 卤素阻燃剂具有相对良好的性能和适 中的价格,尤其是它们与聚合物具有极高的相容性, 但是卤素阻燃剂在应用时会产生大量烟雾和有毒有 腐蚀性气体(溴化阻燃剂,如多溴二苯醚燃烧会产 生溴化二苯并二噁英和溴化二苯并呋喃等,也会释 放具有腐蚀性的卤化氢气体[10]),不利于火灾救援, 其安全性和环保等也备受争议[11].

近年来,较为环保的非卤化产品替代传统卤化 阻燃剂成为一种趋势^[7].磷系阻燃剂凭借低烟、低 毒和高效阻燃效果等已获得市场好评,成为当前研究的热点.磷系阻燃剂应用最广的是磷酸酯和膦酸 酯类,它们资源丰富,价格低廉,与高聚物相容性 好,具有增塑和阻燃双重效用;而膦酸酯类阻燃剂 化学稳定性强,耐水和耐溶剂,阻燃性能持久^[12].有 机磷阻燃剂阻燃原理主要是表面受热时与基质材料 反应产生结构更趋稳定的交联状固体物质或炭化 层,从而阻止聚合物进一步热解的同时隔绝内部热 分解产物继续参与燃烧过程,但是,磷系阻燃剂也存 在挥发性强和遇热稳定性差等缺点^[13].

其他阻燃剂主要为以上阻燃剂的改进或复合, 典型的膨胀型阻燃剂便是一种以氮、磷为主要成分 的新型复合阻燃剂,该类阻燃剂在受热时发泡膨胀 阻燃之作用机制与磷系阻燃剂相同,也存在氮和磷 的协同阻燃的作用,是一种高效低毒环保型阻燃剂, 在纺织品阻燃应用极具潜力^[14].然而,这种膨胀型 阻燃剂与聚合物相容性较差,会导致基质材料的物 理力学性能和电性能降低[10].

此外,也有研究表明,生物大分子,如蛋白质(乳清蛋白、酪蛋白和疏水蛋白等)、多糖和脱氧核糖核酸修饰后可应用于织物,特别是纤维素等棉、聚酯和涤棉混纺物上,已显示出意想不到的阻燃/抑燃特性.有机高分子的阻燃机制可能是其特殊化学成分经加热或暴露于火焰后与基质交互反应,形成稳定和具有保护作用的炭层,进而限制氧气交换和织物挥发可燃性,表现为对纺织阻燃性的有效提高[15].

剩余污泥胞外聚合物(extracellular polymer substance, EPS)是指一种复杂的有机物混合物,其主要成分为多糖和蛋白质等物质,亦可表现出较好的阻燃能力^[5],其将提取自活性污泥与好氧颗粒污泥的两种 EPS 物质喷涂到亚麻织物上亦显示出较好的阻燃特性. 颗粒污泥 EPS 已达到航天飞机内饰物阻燃标准,而普通活性污泥与现有阻燃剂性能不相上下,揭示出 EPS 具有不凡的阻燃特性^[5].

2 EPS 阻燃机制与过程

从上述各种阻燃剂材料特性初步判断, EPS 阻燃机制很可能与其中含磷(P)以及复杂的有机物结构等相关.

2.1 含磷(P)阻燃原理

EPS 物质中含有大量 P 元素,了解和分析其成分与含量将有助于获悉 EPS 阻燃机制与效果.

剩余污泥中 P 来源主要存在两种途径:①生物 聚磷:②化学磷沉淀:前者主要发生于聚磷菌 (PAOs)新陈代谢过程,以聚磷酸盐(poly-P)形式储 存于细胞内,而后者主要沉积在细胞外 EPS 结构 中,EPS中的P含量如见表1所示. 文献[16]采用 扫描电镜结合能谱仪(SEM-EDS)观测生物除磷工 艺(EBPR)活性污泥、发现 EPS 中 P 占污泥总 P (TP)质量的 27%~30%;有研究发现^[17],EBPR 系 统好氧结束段污泥 EPS 中 P 质量为 0.06 ~ 0.09 mg·mg⁻¹(以 EPS 计),占污泥 TP 的 13%;还有研 究采用离子交换树脂法(CER)提取活性污泥 EPS^[18], 测得 P 占污泥 TP 的质量分数为 6.5%~ 10.5%. 然而, Wang 等[19] 用热提取法提取 EPS 后发 现, EPS 中 P 占活性污泥 TP 的质量分数高达 22%~ 47%,好氧颗粒污泥 EPS 中 P 质量分数亦高达 20% ~44%; 更有甚之,龙向宇等[20]采用超声波-阳离子 交换树脂法提取不同来源污泥 EPS,测得其中 P 质 量竟高达污泥 TP的 34%~57%. 对比目前市场磷系 阻燃剂中 20%~30%的 P 质量占比[21], EPS 中的 P 含量刚好处于同一水平.

污泥 EPS 中 P 成分较为复杂,张志超等^[25]利用核磁共振磷谱(³¹P-NMR) 对 EPS 中磷化合物形态进行了分析和定量;按照磷化合物结构可分为正磷酸盐(ortho-P)、焦磷酸盐(pyro-P)、磷单脂(mono-P)、聚磷酸盐(poly-P)和 DNA 磷等 5 种形式,其中,聚磷酸盐(poly-P)又可细分为聚磷末端磷(end poly-P)和聚磷中部磷(middle poly-P);亦可直接将磷化合物分为无机磷和有机磷两大类.表 2 中列出了不同形式 P在 EPS 中的质量占比,可以看出,不同工艺剩余污泥EPS 中 P 含量差别较大;但整体上看,活性污泥EBPR 系统 EPS 中 P 主要以聚磷酸盐形式存在,而传统活性污泥(CAS)系统主要为正磷酸盐^[25].因此,EPS 中的 P 基本上以无机磷形式存在.

表 1 污泥 EPS 中 P 含量

Table 1	Phosphorus	content	contained	in	FPS
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污泥种类	EPS 提取方法	EPS中 P占比/%	文献
EBPR 剩余污泥	离心	27 ~ 30	[16]
EBPR 好氧段末端	CER	13	[17]
活性污泥	CER	6. 6 ~ 10. 5	[18]
活性污泥	CER	6.4 ~ 7.3	[22]
活性污泥	加热法	22 ~ 47	[23]
活性污泥	超声波 + CER	34 ~ 57	[20]
颗粒污泥	甲醛-NaOH	19 ~ 44	[24]
颗粒污泥	加热法	45. 4	[19]
	7. 6	1 1 100	10

表 2 不同来源污泥 EPS 中 P 的存在形式 [25]

Table 2 Different phosphate forms in different activated sludge

磷形态	CAS 污泥 EPS/%	A ² /O 污泥 EPS/%	A ² /O-MBR 污泥 EPS/%
正磷酸盐	58. 97 ± 1. 21	17. 85 ± 1. 49	17. 89 ± 1. 21
焦磷酸盐	23. 38 \pm 1. 31	29. 39 ± 2.02	27.29 ± 2.39
聚磷末端磷	14.98 ± 0.97	7. 45 ± 0.35	8.08 ± 0.24
聚磷中部磷	2.67 ± 2.75	45. 31 ± 2. 22	46.74 ± 1.43

分析磷系阻燃剂阻燃可能机制,可总结为:①生成热稳定物质,形成保护层,降低氧气扩散等过程;②脱水炭化,降低火焰到凝聚相的热传导;③挥发性磷化合物释放减缓燃烧反应.

2.1.1 生成热稳定物质

EPS 含有大量磷酸基团(游离磷酸基团、蛋白质磷酸基团和脱氧核糖核酸磷酸基团等),其作为涂层与亚麻织物等燃烧过程中会转变磷高聚物形式,生成新的沉淀形式,如碳酸羟基磷灰石等^[26].碳酸羟基磷灰石是一种惰性物质,具有极高的热稳定性,它本身就是一种出色的阻燃材料,包裹到有机物表面后可起到很好的阻燃效果^[27],由此推测,碳酸羟基磷灰石很可能是颗粒污泥 EPS 涂层亚麻编织物阻燃特性的主角之一.

在颗粒污泥 EPS 本森(Bunsen)垂直燃烧实验中,红外光谱(FTIR)对残留灰分分析结果表明,蛋

白氨氮基团被全部消耗而无法检出,但能检测到磷酸盐基团和碳酸盐基团存在^[5]. 进一步分析得知,本森垂直燃烧实验温度高达 850°C,否定了碳酸钙的存在(在 825°C温度下 $CaCO_3$ 均被分解为 CaO 和 CO_2);对剩余灰分进行 X 射线荧光衍射(XRD)分析显示,颗粒污泥 EPS 涂层亚麻织物燃烧后确实存在碳化羟基磷灰石 $[Ca_{10}(PO_4)_6(CO_3)]^{[5]}$. 有研究解释了碳酸羟基磷灰石的生成原因:藻酸盐(EPS 主要成分物质)燃烧会导致环境中 CO_2 浓度升高,促进一OH 基团反应而被 CO_3^{2-} 取代,进而生成碳酸羟基磷灰石结构^[28].

但无法否认是否存在其他稳定性物质[如白磷钙石(WHT)、鸟粪石和磷酸铁等]也对 EPS 阻燃特性做出了贡献^[26],因为有研究也发现,镁和磷元素最初共同存在于有机聚合物中,经过高热可能会形成一种新的形式沉淀,但仍需进一步研究^[26].也有研究表明在 EBPR 污泥中证实了上述各种稳定矿物成分存在,当然这也取决于污泥类型,进水特性和运行等条件对污泥 EPS 的生成影响^[6].

然而,分析絮状污泥 EPS 中并未找到碳酸羟磷灰石的存在迹象^[5],这可能是絮状污泥 EPS 阻燃效果较弱于颗粒污泥 EPS 的原因之一. 虽然絮状污泥与颗粒污泥形成过程均存在生物聚磷现象,但是,颗粒污泥内部 pH 值在 9~11 时可能存在碳酸盐和磷酸盐共沉淀,会营造有利于羟基磷灰石生成的环境^[29,30]. 另一方面,颗粒污泥核内网状纤维结构稳定状态更利于羟基磷灰石沉淀过程的发生,而松散的絮状活性污泥显然不利于羟基磷灰石形成^[6]. Mañas 等^[31]在研究中观察到好氧颗粒污泥中存在明显白色结晶沉淀物,其主要成分为羟基磷灰石.

2.1.2 促进脱水炭化

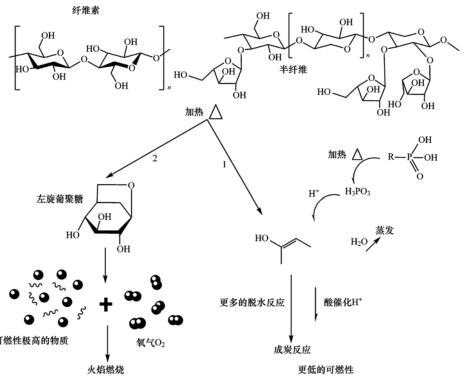
EPS 存在磷酸基团可以促进织物炭化过程,进而减缓燃烧内沿. Kim 等^[5]在燃烧实验中发现了有趣现象,无论是颗粒污泥 EPS 还是絮状污泥 EPS,都表现为直接降低织物分解温度,使织物在较低温度下便开始分解;这种低温环境并未达到织物燃烧温度,意味着织物等有机物只能通过厌氧机制热解(炭化)^[21]而非燃烧,从而间接达到阻燃效果;这种先低温分解后炭化的两步或多步物质分解是炭化体系一个重要基本特征,这种缓慢的炭化分解速度也就保证了完整炭化和良好阻燃效果^[32].

因此,推测 EPS 阻燃机制也与织物炭化作用有 关.以纤维素织物为例,其热解过程存在两种竞争性 反应(如图 1):①纤维素脱水反应炭化;②解聚反 应形成左旋葡聚糖^[21].前者反应产物焦炭在织物表 面会形成致密的高热稳定性泡沫层,可起到绝缘层作用,隔绝热源和气源,阻止内部材料进一步燃烧^[9];后者解聚反应分解虽也会形成一些焦炭,但大部分产物为分子量极低、且可燃性极高的物质,这些物质与氧气反应会维持燃烧和火焰传播过程.显然,从阻燃角度分析,当然希望尽可能促进第一步

反应而抑制第二步反应. EPS 热解释放所形成的磷酸基团,刚好与纤维素表面—OH 基团发生下述式(1)反应,可促进纤维素脱水炭化反应并抑制左旋葡聚糖生成反应.

$$(HPO3)n + Cx(H2O)m \longrightarrow$$

$$["C"]x + (HPO3)n·mH2O$$
(1)





纤维素通过解聚和脱水,在有膦系阻燃剂存在的情况下反应向2个方向进行,最终成炭

图 1 纤维素热解反应[21]

Fig. 1 Pyrolysis process of cellulose

从脱水炭化角度而言,存在磷酸基团和—OH 基团炭化反应即可发生,这意味着 EPS 并不仅仅可 用作纤维素织物涂层材料,也可广泛应用于含氧高 聚物,如沥青、橡胶、皮革和一些塑料等^[12],这就为 EPS 作为表面阻燃剂拓展了应用渠道. 从 EPS 角度 看,无论是絮状污泥 EPS 还是颗粒污泥 EPS 均含有 磷酸基团,均具有较大的阻燃应用潜力.

2.1.3 挥发性磷化合物抑制

有研究表明,磷酸基团除可能作为上述炭化过程催化剂外,还可以直接抑制火焰. EPS 中含磷基团受热分解逐步发生如下变化: EPS(磷化合物)→磷酸→偏磷酸→聚偏磷酸. 燃烧初期磷酸和偏磷酸在聚合物表面覆盖与空气隔绝,高聚物分解完成后形成聚偏磷酸是不易挥发的稳定化合物,具有较强酸性和脱水性,促进成炭^[12].

热分解过程中还会使聚合物受热分解释放出挥 发性磷化合物.质谱分析表明,此时环境氢原子 (H·)浓度会大大降低,说明 PO·可以捕获 H·,即, PO·+H·=HPO^[33]. Sonnier 等^[33]研究某种复合材料(PES/F-MVP),证实燃烧过程挥发性磷化合物释放现象,它与H·和·OH作用可减缓燃烧. 这种阻燃机制与卤素阻燃剂(特别是卤化氢类)非常类似^[32]:含磷化合物在高温下释放的PO·捕获了气相中促进燃烧反应的H·和·OH,阻断或减缓了烃类分支或燃烧链式反应,从而减少热量产生而抑制燃烧,反应生成的水蒸气也降低了表面温度并稀释气相可燃物浓度,具体反应过程见式(2)^[32].

$$PO \cdot + H \cdot \longrightarrow HPO$$

$$PO \cdot + \cdot OH \longrightarrow HPO_{2}$$

$$HPO + H \cdot \longrightarrow H_{2} + PO \cdot$$

$$\cdot OH + H_{2} + PO \cdot r \longrightarrow H_{2}O + HPO \qquad (2)$$

$$HPO_{2} \cdot + H \cdot \longrightarrow H_{2}O + PO$$

$$HPO_{2} \cdot + H \cdot \longrightarrow H_{2} + PO_{2}$$

$$HPO_{2} \cdot + OH \longrightarrow H_{2}O + PO_{2}$$

2.2 类藻酸盐(ALE)阻燃机制

EPS 除传统的含磷阻燃机制外,其核心组分胞外

多糖物质(主要成分为类藻酸盐: alginate, ALE)可以为 EPS 阻燃特性做出较好地贡献. 有研究表明,不同污泥 ALE 含量差异较大,一般介于 10%~40% 挥发性固体(VSS)质量,且颗粒污泥 ALE 含量高于普通活性污泥^[6,34,35],这也是颗粒污泥 EPS 阻燃效果优于活性污泥的重要原因之一. 如图 2 所示,典型 ALE 化学结构呈现纤维长链态,其阻燃机制可归纳为:①炭化阻燃;②基团反应;③凝胶稳定;④生成沉淀.

2.2.1 炭化阻燃

一般认为 ALE 在 300℃左右会发生炭化反应, 热解产生多个中间产物小分子,一部分形成焦炭,另 一部分促进成炭^[37]. 这种 ALE 在燃烧过程中自身较高程度的炭化特性会导致其一旦离开火源,火焰则很快自熄自灭^[38]. 此外, ALE 纤维本身极限氧指数可高达 34%, 意味着其纤维结构并不易熔融,燃烧过程缓慢,拥有良好的热稳定性^[39]. 有研究发现燃烧前后 ALE 总热释放量(THR)和热释放速率(HRR)降低^[40];扫描电镜(SEM)检测发现燃烧后ALE 残基结合紧密,表面孔洞较少;这或许是 ALE 在燃烧过程形成一层具有粘性的残层(主要是炭化物质),阻断火焰和 ALE 物质之间的热传递,实现了较好的阻燃效果.

图 2 藻酸盐(ALE)化学结构^[36]

Fig. 2 Chemical structure of alginate (ALE)

2.2.2 基团反应

ALE 热分解过程大致分 4 步进行: 1060 ~ 170℃ 下,ALE 脱去内部结合水(脱羟基反应); ②220~ 280℃时,物质部分裂解发生脱羧反应; ③300~ 370℃后,中间产物继续裂解、炭化; ④560℃始,氧 化生成 Na,O^[34]. 从 ALE 热分解反应可以看出,ALE 结构存在的大量羧基和羟基在燃烧时可大量吸收空 气中水分,会降低织物表面温度,达到阻燃效果.与 此同时,羟基和羧基遇高温火源时,极易发生酯化反 应脱羟基释放水分,也会伴随着物质表面温度的降 低,且该过程可一定程度上促进炭化反应进行[41]. 在220~280℃期间,物质部分裂解发生脱羧反应, 羧酸盐脱羧生成碳酸盐(如海藻酸钙可转换为碳酸 钙)附着于纤维表面,形成保护屏障[37];同时,ALE 糖苷键也会随机裂解,形成无水糖,裂解产物碎片重 排、脱水、脱羰、脱羧、化学键断裂,缩合反应会产 生不同低分子物质,如糖醛、二氧化碳(CO2)和2-乙酰氧基和2,3-丁二酮等非常稳定的产物[42]. 通过 傅里叶变换红外光谱(FTIR)和质谱联用热重分析 仪(TG-FTIR-MS)分析得知,脱羧和酯化反应为同步 进行: 但从糖醛产率来看,脱羧反应要比酯化反应 发生的概率更大[42]. 无论如何, 这些基团分解和重 组的过程都伴随着大量吸热/放热反应发生,均可降 低织物表面温度. 另外, 酯化反应脱除的 H,O 和脱 羧反应产生的大量 CO₂(占热分解失重的 91.6%) 可进一步稀释可燃气体与氧气浓度[37].

2.2.3 凝胶稳定

ALE 交联阳离子(尤其是二价阳离子)可以结合 ALE 区段中钠离子的结合位点,从而发生交联并形成三维凝胶网络^[43]. 在同时存在二价和三价金属离子时, ALE 会形成坚固、刚性和有序的凝胶结构^[38]. 有研究对 ALE 凝胶热稳定性研究发现,质量分数为3%的 ALE 在0. 18 mol·L⁻¹的 CaCl₂ 和2 mm凝胶直径条件下,凝胶热稳定性最高^[44]. 同时,金属离子也可与糖醛酸残基发生离子作用,形成所谓"鸡蛋盒"结构(见图3)^[42]. 这些凝胶特性可形成无定形团状结构,相互之间极易发生粘连,使得热传递阻力逐渐变大,可间接抑制火焰蔓延^[38]. 进言之,上述结构同时也具有良好亲水性和稳定性,使得 ALE可以较好地黏附在织物表面形成涂层,起到很好的阳燃效果^[6].

2.2.4 生成沉淀

在 ALE 热分解过程中,金属离子会部分沉积在表面并在一定温度范围内起到保护和隔绝的作用. 例如,对于 ALE-Ca 涂层纤维,在 180~350℃会生成 CaCO₃ 与 Ca(OH)₂;海藻酸铝和海藻酸铁分解产生氧化铝和氧化铁^[45]都可以作为保护层附着在纤维表面,从而阻止可燃性气体向外释放,并阻止外部氧气向内扩散,进而抑制纤维素热分解;随 Ca²⁺浓度升高,保护层致密程度会增加,阻燃效果也会相应增加^[37].

显然, EPS 中 ALE 复杂化学结构提高了 EPS 的

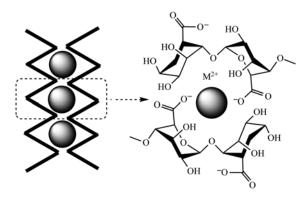


图 3 金属离子与藻酸盐(ALE)形成的"鸡蛋盒"结构^[37]
Fig. 3 The "egg box" structure between metal
ions and alginate (ALE)

阻燃特性. 但是,目前有关 ALE 类物质阻燃机制的研究较少,需要进一步深入研究.

2.3 其他阻燃机制

2.3.1 胞外蛋白阻燃

除了 ALE 这种多糖物质, EPS 中发现的胞外蛋白(PN)亦能表现出良好阻燃性. 淀粉样蛋白质物质化学和热稳定性都很强,即使在强变性试剂(2 $\text{mol}\cdot\text{L}^{-1}$ 硫脲+8 $\text{mol}\cdot\text{L}^{-1}$ 尿素+3% SDS)中煮沸60min也只会部分分解^[46].

由于传统温和提取手段无法分离这种糖蛋白, 所以,人们一直没有注意到此类物质为 EPS 贡献相 当的结构强度和刚性.有研究表明,糖基化似乎使蛋 白质更具有化学稳定性和耐热性^[46],这意味着它有 着较强的阻燃性能,但其阻燃效果与过程机制还有 待进一步研究.

2.3.2 协同阻燃

EPS 阻燃潜质也得益于 EPS 自身协同阻燃,这种"1+1>2"的协同作用主要发生在磷与氮之间以及磷与 ALE 之间.

(1)磷与氮协同阻燃. 磷阻燃过程主要发生在物质固相表面;若引入另一阻燃元素实现其在气相中阻燃将可能实现更为高效的阻燃效果. EPS 中存在较高含量氮(N)元素(源于蛋白质氨基酸等基团)在高温受热状态较易释放 CO₂、N₂、NH₃和H₂O等不可燃气体,这些不可燃惰性气体可稀释环境中氧气以及聚合物受热分解产生的可燃物浓度;同时,热对流过程也会带走一部分热量,减缓燃烧;甚至含N链段与含P基团会生成含有P—N键中间体,是良好的磷酸化试剂,更易促进磷酸化反应,从而增强上述含磷炭化阻燃作用;此外,含N化合物还可以延缓凝聚相中磷化合物的挥发损失,加强P的氧化,放出更多惰性气体,提高阻燃性能^[9]. 以酪蛋白为例,它是胶束结构中含有大量磷酸盐基团的聚氨基酸,其分解过程与聚磷酸铵相似,产生聚磷酸

和氨,这两种物质协同可发挥更好的阻燃效果.

(2)值得一提的是 ALE 对于溶解态 P 具有很强的吸附作用,大量的磷酸盐可以沉积于这种多糖类物质表面^[47],从而进一步强化上述的含 P 阻燃过程.

3 EPS 阻燃效果评价

上述分析揭示了 EPS 具有复杂的化学结构与性质,推测 EPS 作为阻燃剂潜力很大. 但是,如何判定和评价 EPS 阻燃效果与特性,并将之与目前商用阻燃剂优势比较也是研发工作需要阐明的问题. 根据目前研究结果,以下总结 EPS 与目前商用阻燃剂性能对比.

3.1 阻燃表征

现有研究较少直接表征 EPS 阻燃特性,但可以借鉴其他材料阻燃表征方式评价 EPS 燃烧行为.

极限氧指数法(LOI)是表征材料燃烧行为的指数,可以使用氧指数测定仪测定 $[^{48}]$. 常用标准有欧洲 ISO-4589、日本 JISK-7201 以及中国 GB/T 5454-1997 $[^{49}]$.

垂直燃烧法适用于有阻燃要求的服装织物、装饰织物和帐篷织物等阻燃性能的测定,是现如今最为普遍常用的测试方法^[50]. 我国一般按照《纺织织物-阻燃性能测定-垂直法》(GB 5455-97)方法对EPS 阻燃性能进行测试. 另外,还有水平燃烧法、45℃燃烧法和 UL94 燃烧法等,它们的原理与结果大同小异^[51].

此外,燃烧过程中 CO 释放量以及烟释放量 (TSR)等指标也都是阻燃材料安全使用性能的重要 衡量标准,可以进行如发烟性实验和防护性能实验 等^[51].

EPS与表面基质材料的结合行为也需要在微观层面研究分析,有助于进一步理解、验证 EPS 热解和燃烧特性.用扫描电子显微镜(SEM)可观察 EPS与表面材料结合情况,包括结合点位置和结合方式等,以分析涂层修饰手段的有效性^[21,33].采用热解燃烧流动量热法(PCFC)可描绘 EPS 热释放速率曲线(HRR)和峰值热释放速率(pHRR),进而研究样品在微观尺度下的燃烧行为;曲线一般出现两个峰值,即初始最高峰和熄灭前另一个高峰,前者代表材料燃烧时炭化形成的炭层,它可阻隔、减弱热向材料燃烧时炭化形成的炭层,它可阻隔、减弱热向材料内层传递,目的是表征材料的成炭性能^[48];采用热重分析法(TGA)描绘 EPS 热失重曲线可研究其热稳定性^[21],并且可以直观分析判断可燃性物质挥发速率.使用红外光谱仪(FTIR)可分析燃烧实验前后涂层材料,分析不同波峰代表剩余残炭中的官能

团,进而厘清 EPS 参与燃烧反应的内含物或基团[5].

3.2 阳燃特性

现有直接描述 EPS 阻燃特性的实验研究还很少. Kim 等^[5]把提取自活性污泥与颗粒污泥的两种 EPS 物质按照 EPS 水溶液 3% (质量分数)比例喷涂到亚麻织物上,风干 72 h 后进行 Bunsen 垂直燃烧实验,以研究涂层亚麻织物的燃烧特性,结果显示,活性污泥 EPS 和颗粒污泥 EPS 涂层亚麻织物的灭火和熔体滴落时间均为 0 s,烧焦长度分别为 260 mm 和 130 mm. 美国联邦航空条例中运输类飞机座舱内部阻燃材料阻燃性能要求灭火时间和熔体滴落时间应分别 <15 s 和 <3 s,燃烧长度应 <152.4 mm (6 英寸)(US-FAR 25.853 标准),污泥 EPS 灭火时间和熔体滴落时间参数显然符合飞机内饰材料标准,但烧焦长度仅颗粒污泥 EPS 符合要求.这一阻燃特性实验结果初步证明,EPS 确实具有较好阻燃特性.

上述实验研究进一步观察织物燃烧后,发现均存在放热现象,在两种 EPS 涂层亚麻织物均出现了红光(阴燃)现象,且颗粒污泥 EPS 涂层亚麻织物阴燃时间非常短,大约是絮状污泥 EPS 的 1/4. 结果说明,颗粒污泥 EPS 涂层对热量传递起到了屏蔽作用;进一步细致观察发现,絮状污泥 EPS 涂层亚麻织物阴燃过程沿表面星现出一定的路径,这可能是因为活性污泥 EPS 结构较颗粒污泥松散所出现的阴燃裂缝^[5].

总之,颗粒污泥 EPS 阻燃效果明显优于絮状污泥,这可能是颗粒污泥微生物自我聚集形成了具有良好的包裹性、黏附性和延展性结构所致;颗粒污泥 EPS 这种网状纤维基质(纤维直径约为 40~50 nm)^[5]错综缠绕更有利于形成连续层状涂层覆盖于织物表面,导致较好的阻燃特性. 然而,絮状污泥 EPS 是块状拼接纤维聚集而成,表面可观测到细微裂缝,块状直径约 80~100 nm,其裂缝长度达300 000 nm,宽度可达 45~50 nm^[5],所以,较难形成颗粒污泥状表面连续层状结构,涂层到织物表面后会出现裂缝间隙,放热阴燃可以发生在这些间隙中,从而影响絮状污泥 EPS 阻燃效果.

对市场上热门喷涂型阻燃材料——聚氨酯泡沫燃烧测试(采用 KS-5004 建筑材料水平燃烧测试仪)结果显示,其平均燃烧时间为 10 s,燃烧长度为130 mm,似乎还略逊于 EPS^[52]. 这意味着颗粒污泥 EPS 具有较大的阻燃市场潜力.

3.3 潜力分析

从以上 EPS 物质阻燃特性机制归纳可知, EPS

物质阻燃不仅取决于 P 含量多寡,其它成分与复杂结构(藻酸盐类、淀粉样蛋白等) 也让 EPS 阻燃性能别具一格,使其具有较大应用潜力. 对比市场阻燃剂材料, EPS 具有以下 7 大优势.

- (1)与氢氧化铝等金属阻燃剂比较, EPS 阻燃 材料来源于微生物代谢, 具有绿色环保、易降解优势, 阻燃效果好, 相对用量减少, 不会影响材料本身 物理机械性能; 同时, 经处理后的 EPS 溶液本身无 色无味, 扩大了修饰其他材料的应用空间和加工 特性.
- (2)与其他含卤阻燃剂或磷系阻燃剂对比,EPS 磷物质分子量较高(如聚磷酸盐等),具有不易挥 发、热稳定性好、且不产生有毒气体等特点,可弥 补某些阻燃剂燃烧烟雾大的缺陷.
- (3) EPS 除含有大量 P 元素外,还含有大量蛋白质类物质,这些物质在燃烧过程中可实现较好的 N-P 协同阻燃效果,其作用远高于化工材料单一组分合成所产生的阻燃作用^[9].
- (4) EPS 含有大量天然多糖高分子(如 ALE 等),具有多羟基碳链结构,代替纤维织物直接炭化脱水形成焦炭,与磷基团脱水炭化作用协同,比单一磷系阻燃剂或卤素阻燃剂效果更佳^[53].
- (5) EPS 作为阻燃剂表面涂层材料具有优良的耐水性(防水性),可增加涂层材料使用寿命.目前市场含磷阻燃剂(如 poly-VPA)涂层表面干附着力很好,但水浸实验后附着力较差,阻燃剂成分极易被冲刷.工程上一般采用与疏水性单体共聚方式来提高涂层湿附着力,但是以降低物质一定防火性能为代价^[54].而 EPS 中结构性糖类物质(ALE 等)既有亲水性又有疏水性;亲水性多糖端会牢固地附着于纤维表面,而疏水性球状结构脂类则朝外衍生;水滴浸入后疏水基团斥力会使水分子保持水滴状而起到耐水性能,防止冲刷^[6].
- (6) EPS 中存在"交联剂"成分,蛋白结构上多种官能团可提供更多与阳离子和其他有机物结合位点和作用点,增加了与更多材料相容性可能;多糖和淀粉样蛋白等的水凝胶特性在 EPS 与表面材料接枝时发挥交联稳固作用,可增强 EPS 物质吸附粘合能力^[6,46].
- (7) EPS 为微生物合成有机成分,作为阻燃剂使用流入自然环境很容易被生物降解,这一点是市场现有阻燃剂(如卤代等无机阻燃剂)无法比拟的.因此,EPS 作为阻燃剂的环保性毋容置疑.

然而,也正是因为 EPS 独特结构特征,某些情况下并不适用于特定金属或塑料等材料表面阻燃修饰,且现阶段 EPS 提取、回收和应用技术距离实现

其完全替代其他阻燃剂还有一定差距.

4 EPS 阻燃性能提升策略

EPS 虽具有优于市场阻燃剂之潜力,但从应用经济性角度考虑应进一步探究阻燃能力提升方式,以降低获取的经济成本. 从机制出发,阻燃性提升一方面可以通过提高 EPS 中 P 含量,亦可考虑提高和纯化 EPS 中与阻燃相关的物质成分(如 ALE 和淀粉样蛋白质等). 另一方面,需要考虑 EPS 与基质织物结合方式,尽可能保证在基质性能基础上提高修饰物整体阻燃性能.

4.1 富磷 EPS 提取

有研究表明,物质中 P 含量越高,热速率释放峰值(pHRR)便会降低,而热解后焦炭量将会增加^[21]. 这意味着 EPS 含 P 量越高,其阻燃效果可能越好.

不同提取方法将导致 EPS 提取中 P 含量出现 差异. 将实验室培养 EBPR 污泥分别采用甲醛/ NaOH、超声、EDTA、加热和阳离子交换树脂 (CER)等5种方法与直接离心方式进行比对,分别 表征所提取到的 EPS 中 P 含量. CER 法在加入 70 g·g⁻¹(以 CER/VSS 计),并在 500~600 r·min⁻¹离 心转速下提取 6 h, 获得的 EPS 中 P 含量约占污泥 TP 的质量分数为 6.6%~10.5%; 而其他提取方法 EPS中P含量相对较低或者完全破坏细胞壁,导致 EPS 提取"失效"[18],各种方法提取结果示于图 4. 不同形态污泥 P 含量也不尽相同,有研究表明,活 性污泥 EPS 含 P 质量(以 EPS 计,下同)为 0.09~ 0.35 mg·mg⁻¹, 高于颗粒污泥的 0.024 ~ 0.071 mg·mg⁻¹,这可能是因为颗粒污泥的 EPS 产量远高 于活性污泥,导致单位 EPS 中 P 含量降低; 因此颗 粒污泥中 TP 累积量(TP_{EPS}/TP_{Sludge}高达 45.4%)普 遍高于活性污泥^[55].

此外,其他诸如 C: N: P^[56]、进水基质^[34,57]、

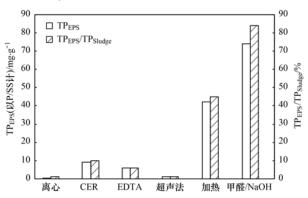


图 **4** EPS 不同提取方法中 P 含量对比^[18]

Fig. 4 Comparison of P content in EPS extracted $\qquad \qquad \text{by different methods}$

 $pH^{[4]}$ 和温度 $^{[17]}$ 等都会影响污泥中 EPS 含量,进而可能影响其中 P 的本底含量.

4.2 ALE 纯化与提高

阻燃剂的阻燃性能不仅仅与 P 含量有关,还与 其热稳定性和阻燃剂核心分子在织物纤维体中通过 性等相关^[21,33]. ALE 是 EPS 主要有机物阻燃成分, 因此,了解 ALE 生成影响因素等环境条件将有助于 提高 EPS 中 ALE 含量,进而提高污泥 EPS 的阻燃 潜力.

污水处理工艺运行条件(如碳源、有机负荷、 溶解氧和 pH) 会直接影响污泥中 ALE 合成[58]. 碳 源不同导致污泥中菌群结构发生改变,进而影响 ALE 分泌, ALE 理化性质和分子结构也会产生差 异[59]. 吴振宇等[60]的研究发现,以葡萄糖基质培养 微生物所产生的胞外多糖量(ALE 为多糖高聚分 子,以 MLSS 计,下同)为 6.3 mg·g-1,而以蛋白胨培 养的微生物产生的胞外多糖量是 4.9 mg·g-1. 也有 实验发现,有机负荷也会影响污泥 ALE 分泌; 当有 机负荷突然增加会刺激颗粒污泥分泌更多的环二鸟 苷酸(c-di-GMP,一种促进颗粒污泥形成的物质,调 节纤维素形成的信使分子[61]),从而促进 ALE 进 步分泌[62].溶解氧(DO)也会影响微生物代谢,从而 导致 ALE 产量不同. 有研究发现, DO 升高后 ALE 产量会增加[63]. 也有研究表明[64], pH 也会影响 ALE 合成;酸性条件下, EPS 中多糖质量下降约 30%, 意味着 ALE 含量也下降; 而强碱条件下, 胞外 多糖提取质量升高约 15%,导致 ALE 含量相应

不同污泥形态也会影响污泥 ALE 合成与结构. 有研究发现,颗粒污泥中 ALE 提取量[160 mg·g⁻¹,以 MLSS 计,下同]是絮体污泥提取量(72 mg·g⁻¹)的 2 倍多,而且二者 ALE 中古罗糖醛酸与甘露糖醛酸含量比值分别为 8:1和 2:1,表明不同形态污泥 ALE 结构确实有所不同^[35].

另一方面,不同提取方法也会导致 EPS 中 ALE 含量和性质变化,最早借鉴大型海藻(如海带)提取藻酸盐的方法有:钙凝离子交换法、钙凝酸化法、酸凝酸化法和酶解法^[65,66]. 随后,有研究首次借鉴酸凝酸化法采用高温碳酸钠以及酸析出和有机萃取方法成功从颗粒污泥中提取出 ALE 物质,其提取量高达(35.1±1.9)%(以 SS 计)^[67]; 再后来,文献[6,34,35]继续深入研究和优化方法,增加提取物纯度,试图提高 ALE 物质作为表面涂料/防火涂层的性能^[5,6].

不同的提取手法会影响所提取 EPS 结构的完整性以及 EPS 内具有阻燃功能的物质含量. 对前者

完整性保证可能导致后者含量降低; EPS 在阻燃材料应用方面,二者对阻燃特性正向效益影响孰轻孰重,还有待进一步实验研究.

4.3 修饰手段

物质阻燃效果还不仅取决于自身阻燃特性,亦受限于与基质的结合方式,而结合方式直接决定阻燃基团或分子在燃烧过程中阻燃潜力发挥.有研究在 EPS 阻燃实验中直接采用3%(质量分数)的 EPS水溶液,分3次喷涂到织物表面后风干72 h 成型;该方法简易但展现出不俗的阻燃性能^[5].若进一步优化 EPS 修饰织物方法,采用其他接枝(graft)方法,是否可实现更强的阻燃性能?这有待于进一步实验验证.

目前存在许多工艺可使阻燃剂与表面材料结合,主要分为物理法和化学法. 化学法是把纤维织物在 70~160℃下浸泡于含磷阻燃剂溶液(质量分数为 5%~10%)中,约 2 h 左右成型^[21]. 物理法包括UV 照射、等离子体处理、溶胶-凝胶处理、伽玛或电子束照射等^[54,68~70],其中,较为有效的是辐射法,将织物浸入含磷阻燃剂溶液中,利用电子束辐射以不同剂量辐照使二者相互结合,最后水洗去除未结合分子物质^[21];辐射法主要是将磷酸基团被共价结合或聚合后"困"于纤维结构,能在不破坏织物基础上结合足量磷(质量分数为 1.4%),从而提高阻燃性能.

合适溶剂参与可以控制阻燃剂在纤维素中渗透量,从而控制织物涂层材料最终 P 含量. 通过扫描

电镜-能谱分析联用(SEM-EDX)接枝溶剂对纤维修饰的影响,有研究发现甲基膦酸酯(MAPC1)只会导致 P 存在于纤维表面,而并未渗透到纤维体结构之中,导致纤维素中心缺少 P 元素;相反,乙烯基磷酸二甲酯(MVP)使 P 元素分布于纤维素表面至内部纤维块^[33]. 这意味着,选择合适溶剂对 EPS 与织物的接枝过程将可能直接影响最终 P 含量,而限制阻燃特性^[33].

针对 ALE 特定涂层方法,存在钡、镍和钴离子 交联织物涂层手段,具体过程如图 5 所示[71]. ALE 作为一种阴离子聚电解质,与阳离子聚电解质 (PEI)通过逐层组装方法(LBL)[72]在棉织物表面形 成涂层,该过程可以增强棉织物的稳定性;随后利 用该方法在织物表面构建含P聚电解质而具有独 特优势,利用钡、镍和钴离子交联 ALE 对织物进行 涂层. 为防止沉积, 对棉织物水洗后室温下风干过 夜; 然后将底物分别在 PEI 和 ALE 溶液中交替浸泡 2 min,去除未结合化合物. 这种两相交替浸泡可形 成 PEI 和 ALE 双层膜结构. 交替浸泡 10次(形成 10 层双膜结构) 后置于 5 mol·L⁻¹ BaCl₂、C₄H₆O₄Ni· 4H₂O 或 C₄H₆O₄Co·4H₂O 水溶液中浸泡 2h,水洗去 除未反应的金属离子后烘箱 70℃干燥过夜,然后在 干燥器吸收/去除剩余水分[71]. 该方法制备的阻燃 织物燃烧效率明显比未处理织物要低,火焰蔓延率 也要低 28%,燃烧实验中也明显观测到成炭效应,特 别是测得 ALE 与钡离子交联的涂层织物拥有相当 的阻燃性能.

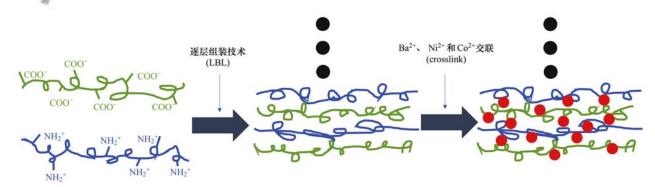


图 5 逐层组装方法(LBL)与钡、镍和钴离子交联藻酸盐(ALE)对棉织物涂层方法[71]

Fig. 5 Schematic construction of alginate-based coatings and crosslinking of coatings by metal ions on cotton fabrics

5 结论与展望

(1)剩余污泥 EPS 作为高附加经济价值物质提取将推动污水处理资源化目标. EPS 具有得天独厚的阻燃物质形成环境,其阻燃性能更是好过目前市场普遍销售的阻燃剂,特别是其可生物降解性是其它阻燃剂无法比拟的环保特性. EPS 中所含有的磷(P)占污泥总磷(TP)质量的 10%~45%,且其形态

从磷酸盐、焦磷酸盐和磷单脂一直延伸至聚磷酸盐 (poly-P)和 DNA 磷等,正是这种含磷特性使其具有 生成热稳定物质、促进脱水炭化和发挥气相磷阻燃 特性.进言之,EPS 中所含类藻酸盐(ALE)物质的复杂化学结构(凝胶态无定形团状结构)和丰富官能 团(羧基、羟基和羰基等)在燃烧或热分解情况下炭化、脱水以及发生热稳定反应时可极大增强 EPS 自身阻燃性能.此外,EPS 其他结构,如蛋白质交联和

氮磷协同等,在从侧面提高阻燃性能方面也都优于市售其他阻燃剂. EPS 所有这些独有特性均为其作为高性能表面涂层防火材料奠定了应用基础.

- (2)通过筛选污泥和选取合适的提取方法可以 提高 EPS 磷含量以及功能性物质含量,进一步提高 其阻燃性能和修饰性能,实现更高性能 EPS 阻燃物 质回收.同时,在此基础上挖掘适宜的阻燃材料与被 保护基质的结合方法,以扩大 EPS 高效防火涂层的 应用范围.
- (3)事实上, EPS 作为表面涂层不仅可用作阻燃材料,亦可用作防水材料. 多糖和蛋白质分子富含羟基和羧基等亲水官能团可增强 EPS 物质与织物的结合性能,它们所具有的相对较低表面电荷和较高整体疏水性能可增强织物的疏水性能. EPS 物质这种双性特征也能使其作为有发展前景的环保防水材料. 而有关 EPS 在食品、医药、纺织、印染、造纸和日化等行业中的应用亦有可能解决藻酸盐物质传统上只能依靠从大型海藻中提取的尴尬处境.

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