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### 批和 HUANJING KEXUE)

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### 含藻水中壬基酚的光降解转化研究

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摘要:选取壬基酚(nonylphenol, NP)作为研究对象,研究了水中2种常见淡水藻(小球藻和鱼腥藻)对壬基酚的光降解促进作用,并对藻引发水中壬基酚的光降解途径进行了分析;同时也研究了天然水中普遍存在的物质成分(腐殖酸和铁离子)与藻协同引发壬基酚的光降解及其影响机制.结果表明,含淡水藻的水溶液经过光照后能引发其中壬基酚的光降解;藻、腐殖酸和铁离子的水溶液经过光照后,对壬基酚光降解的增强促进作用更大,在含藻、腐殖酸和铁离子的水溶液中,4h光照后壬基酚的降解率可达58%.根据此结果推测藻/腐殖酸和铁离子体系光照后能产生更多的活性物质,从而促进水中有机污染物的光降解.

关键词: 壬基酚; 光照; 藻; 铁离子; 降解

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# Study on the Degradation and Transformation of Nonylphenol in Water Containing Algae

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**Abstract:** The photodegradation of nonylphenol induced by two common freshwater algae was investigated. The mechanism of nonylphenol photodegradation induced by algae was analyzed. The synergistic induction of nonylphenol degradation by algae and substances in water such as humic acid and ferric ions was also investigated. Results showed that the algae could induce the photodegradation of nonylphenol. The degradation of nonylphenol in water in the presence of algae, humic acid and ferric ions was obvious and the efficiency of degradation could reach 58% after 4 h illumination. Based on the results, it was speculated that the algae, humic acid and ferric ions system could produce more active oxygen after illumination, which could promote the photodegradation of the organic contaminants in water.

Key words: nonylphenol; illumination; algae; ferric ions; degradation

随着表面活性剂的广泛应用,表面活性剂烷基酚聚氧乙烯醚通过不同途径排入水体,经过污水处理厂的生物化学过程后可转化为壬基酚和辛基酚<sup>[1,2]</sup>.研究表明,壬基酚聚氧乙烯醚生物降解后的产物比其母体毒性更大<sup>[3]</sup>,对人类和其他生物具有内分泌干扰活性和毒性<sup>[4,5]</sup>,且此类物质在环境中降解缓慢,其累积后可达到显示毒性效应的浓度,因此,研究此类物质在环境中的降解与归趋对于人类健康与环境安全具有重要意义.

在天然水中普遍存在着各种生物或非生物的物质成分,例如藻、腐殖酸、铁离子等,太阳光可使这些物质发生光化学过程,从而产生各种活性物质,这些活性物质可引发水体中有机污染物的光降解. 2003 年 Liu 等<sup>[6,7]</sup>研究表明在高压汞灯或紫外灯光照条件下,藻可引发水中持久性有机物的光降解. 2004 年葛利云等<sup>[8]</sup>研究表明在高压汞灯光照下,普

通小球藻可引发水中苯胺的光降解. Conte 等<sup>[9]</sup>曾研究应用腐殖酸作为一种天然的表面活性剂来修复处理高污染的土壤. Fukushima 等<sup>[10]</sup>研究了水中腐殖酸与金属离子对苯胺污染物降解转化的影响. 吴峰<sup>[11]</sup>的研究表明,铁-草酸盐配合物体系在光照作用下能产生氢氧自由基并能促进有机污染物的光降解. Cho 等<sup>[12]</sup>研究表明,铁离子可与非离子型表面活性剂形成复合物,在可见光照射下可使 CCl<sub>4</sub> 脱氯. Chen 等<sup>[13]</sup>研究表明,铁-草酸盐配合物可引发农药草甘膦的光降解.

研究表明,天然水中腐殖酸、金属离子可通过相互作用形成稳定的复合物[14],在天然环境中金属

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离子主要以腐殖质的配合物形式存在[15,16].由于 藻同腐殖酸有某些类似的结构特征,因此可以推测, 藻与金属离子之间自然也会存在类似的复合作用. 藻类、金属离子和腐殖质等可通过相互作用形成复合物,在光照条件下发生一系列的光化学过程,引发水体中有机污染物的光降解[17].由于生物和非生物物质复合物的光作用促进机制的复杂性,到目前为止,对于这些复合物的光促进过程引发的有机污染物的光降解和机制的研究仍不够.本实验主要研究了水中淡水藻类、以及水中藻与铁离子、腐殖质等的复合物(藻/腐殖酸/铁离子)在光的作用下引发壬基酚的降解转化,分别分析了底物的初始浓度、藻的浓度、溶液的 pH 值、铁离子及腐殖酸浓度变化等因素的影响作用.

#### 1 材料与方法

#### 1.1 材料与仪器

#### 1.1.1 试剂

壬基酚(nonylphenol, NP)(分析纯,美国 Sigma 公司),乙腈(分析纯,上海凌峰化学试剂公司),甲醇(色谱纯,美国 Fisher 公司),HCl、NaOH(分析纯),FeCl<sub>3</sub>·6H<sub>2</sub>O(分析纯),腐殖酸(humic acid/HA, Aldrich 公司/USA),水生 4号(HB-4)培养基和 SE 培养基所需试剂. 实验所用水为去离子水. 以乙腈为助溶剂,配制 200 mg·L<sup>-1</sup>壬基酚(体积分数 70% 乙腈)储备液. 腐殖酸储备液为 200 mg·L<sup>-1</sup>(微量碱助溶,0.45  $\mu$ m 微孔滤膜过滤).

#### 1.1.2 仪器和反应装置

高效液相色谱仪:岛津 LC-10ATVP 泵、KR100-5C18 色谱柱(4.6 mm×150 mm,5 μm)、Waters 481 紫外检测器(Waters,美国)、N3000 色谱工作站(浙江大学),TES-1332 数字式照度计(台湾泰仕电子股份有限公司),LD5-2A 离心机(北京医用离心机厂),pH320-S 精密数字酸度计(梅特勒公司),电子天平 AY120(日本岛津),恒温光照培养箱(HGP300型,中国科学院武汉科学仪器厂),显微镜,金属卤化物灯(250 W,波长 365 nm,光强197 800 lx;循环冷却水套外光强为150 000 lx),自制光反应器,自制灯外罩循环冷却水套.

#### 1.1.3 藻的培养与处理

实验中选用 2 种淡水藻: 鱼腥藻 (Anabaena cylindrical, AC)和小球藻 (Chlorella vulgaris, CV). 其中小球藻的培养基组成成分为: 200 mg·L<sup>-1</sup> (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>, 30 mg·L<sup>-1</sup> [Ca (H<sub>2</sub>PO<sub>4</sub>)<sub>2</sub>·2H<sub>2</sub>O +

 $CaSO_4 \cdot H_2O$ ], 80  $mg \cdot L^{-1}$   $MgSO_4 \cdot 7H_2O$ , 100  $mg \cdot L^{-1}$   $NaHCO_3$ , 25  $mg \cdot L^{-1}$  KC1, 0. 150  $mL \cdot L^{-1}$   $FeCl_3$ (1%), 2. 86  $mg \cdot L^{-1}$   $H_3BO_3$ , 1. 81  $mg \cdot L^{-1}$   $MnCl_2 \cdot 4H_2O$ , 0. 222  $mg \cdot L^{-1}$   $ZnSO_4 \cdot 7H_2O$ , 0. 039  $mg \cdot L^{-1}$  ( $NH_4$ ) $_6Mo_7O_{24} \cdot 6H_2O$ , 0. 079  $mg \cdot L^{-1}$   $CuSO_4 \cdot 5H_2O$ , 0. 5  $mL \cdot L^{-1} \pm$   $\pm$   $\pm$ 

鱼腥藻的培养基组成成分为: 250 mg·L<sup>-1</sup> NaNO<sub>3</sub>, 75 mg·L<sup>-1</sup> K<sub>2</sub>HPO<sub>4</sub>, 75 mg·L<sup>-1</sup> MgSO<sub>4</sub> · 7H<sub>2</sub>O,19 mg·L<sup>-1</sup> CaCl<sub>2</sub> · 2H<sub>2</sub>O,175 mg·L<sup>-1</sup> KH<sub>2</sub>PO<sub>4</sub>, 25 mg·L<sup>-1</sup> NaCl, 0.662 mg·L<sup>-1</sup> FeCl<sub>3</sub>, 2 mg·L<sup>-1</sup> Na<sub>2</sub>EDTA,2.86 mg·L<sup>-1</sup> H<sub>3</sub>BO<sub>3</sub>,1.81 mg·L<sup>-1</sup> MnCl<sub>2</sub>·4H<sub>2</sub>O,0.222 mg·L<sup>-1</sup> ZnSO<sub>4</sub>·7H<sub>2</sub>O,0.039 mg·L<sup>-1</sup> (NH<sub>4</sub>)<sub>6</sub>Mo<sub>7</sub>O<sub>24</sub>·6H<sub>2</sub>O,0.079 mg·L<sup>-1</sup> CuSO<sub>4</sub>·5H<sub>2</sub>O,4 mg·L<sup>-1</sup> 土壤浸出液.培养基用 0.1 mol·L<sup>-1</sup>Na<sub>2</sub>CO<sub>3</sub> 调至 pH 7.0 ~ 7.2.

在无菌条件下进行接种,放入恒温光照培养箱培养,培养温度25℃. 培养至藻对数生长期及藻浓度较大时(10~14 d),取出离心(转速4000 r·min<sup>-1</sup>,时间20 min)收集藻体,并用去离子水离心清洗3次,在显微镜下用血球计数板计藻细胞个数(个体计数法),该藻备用.

#### 1.2 实验方法

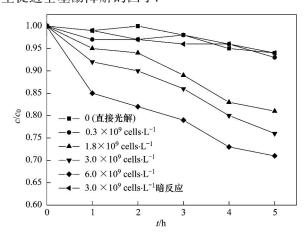
含藥或藻、腐殖酸和铁离子水溶液中壬基酚的光降解实验:用清洗后备用的藻和腐殖酸、铁离子与酚类物质的储备液分别配制一定浓度的反应液,用 HCl 和 NaOH 调节到指定的 pH 值定容后,将其转入分布在金属卤化物灯周围的壬基酚光反应器中,开启金属卤化物灯进行光照反应,同时打开灯的冷却水套水系统,使灯水套周围环境温度维持常温.间隔一定时间(0、1、2、3、4、5h)取样,光照0~5h的样品都在5h后统一离心(3500 r·min<sup>-1</sup>,22 min),取离心后的上清液采用 HPLC 方法分析底物浓度.暗反应同时进行,数据为3次实验的平均值,误差低于5%.

HPLC 的主要测定条件(参数)为:280 nm 波长下检测. 检测 NP 的流动相:甲醇/水:93/7(体积比),流速均为1.0 mL·min<sup>-1</sup>. 在此分析条件下,所用去离子水中没有烷基酚类背景物被检出.

#### 2 结果与讨论

- 2.1 藻引发水中壬基酚的光降解
- 2.1.1 光照和藻的存在对壬基酚降解的影响

为考察光照和藻类同时存在对壬基酚降解转化的影响,本实验选择了模拟自然光的人工光源,选取了天然水中常见的小球藻作为媒介物质,在壬基酚的初始浓度为6 mg·L<sup>-1</sup>,小球藻的浓度范围为0.3×10°~6×10°cells·L<sup>-1</sup>时研究了壬基酚的降解率,实验结果如图1 所示. 有光照时,藻对壬基酚的降解促进作用明显,随着小球藻浓度的增加,壬基酚的光降解率升高. 无光照时,含藻浓度为3×10°cells·L<sup>-1</sup>的壬基酚水溶液中,壬基酚几乎无降解反应,此结果说明,光照对于壬基酚的降解起主要诱导作用. 另外,当无藻时,在光照条件下,壬基酚也几乎没有直接光解,此结果说明,淡水藻在光照后会产生促进壬基酚降解的因子.



 $[NP]_0 = 6 \text{ mg} \cdot L^{-1}, \text{ pH} = 6.5 \pm 0.1$ 

#### 图 1 含小球藻水溶液中小球藻(CV)浓度变化 对壬基酚的光降解影响

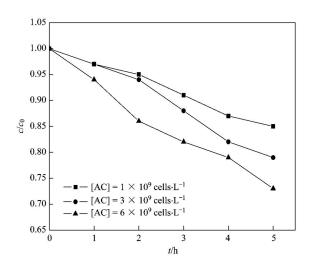
Fig. 1  $\,$  Effect of the change in CV concentration on the photodegradation of NP

#### 2.1.2 鱼腥藻浓度变化的影响

为考察其他淡水藻类在光照时对壬基酚降解转化的影响,本实验选取了另一种水中常见的淡水藻鱼腥藻. 当壬基酚的初始浓度为 6 mg·L<sup>-1</sup>,鱼腥藻的浓度为 1×10°~6×10° cells·L<sup>-1</sup>时,鱼腥藻对壬基酚光降解的影响如图 2 所示. 有光照时,鱼腥藻对壬基酚的降解促进作用也很明显. 随着鱼腥藻浓度的增加,壬基酚的光降解率升高. 此结果进一步说明,水环境中常见的淡水藻类在光照后会产生某种促进壬基酚降解的因子.

#### 2.1.3 藻引发水中壬基酚的光降解途径分析

前期研究表明,藻对壬基酚的吸附较大,且吸附在 50 min 左右达到平衡. 因此,光降解实验时,0~5 h 的光照样品都在 5 h 后同时离心,采用色谱仪分析离心后的上清液中壬基酚的浓度,以扣除吸附量,在



 $[NP]_0 = 6 \text{ mg} \cdot L^{-1}, \text{ pH} = 6.5 \pm 0.1$ 

#### 图 2 含鱼腥藻水溶液中鱼腥藻(AC)浓度变化 对壬基酚的光降解影响

Fig. 2 Effect of the change in AC concentration on the photodegradation of NP

扣除吸附量后的壬基酚降解研究表明,光照后,两种淡水藻对壬基酚的降解有很大的促进作用,进一步分析藻促进壬基酚光降解的原因主要有以下3种情况.

(1) 藻光照后分泌的物质或部分破碎后释放的物质会产生氢氧自由基和其他活性氧物质<sup>[18]</sup>,从而促进壬基酚的光降解,其产生途径分析如下.

$$egin{align*} egin{align*} A \& b & \xrightarrow{hv} & \text{OH} \\ & & & \text{H}_2\text{O}_2 \\ & & & \text{HO}_2 & \\ & & & & \text{HO}_2 & \end{aligned}$$

(2) 藻光照后吸收的光能量转给藻表面的壬基酚,促进壬基酚直接光降解:

$$CV/AC \xrightarrow{hv} {}^{1}CV/AC^{*} \xrightarrow{hv} {}^{3}CV/AC^{*}$$

$$^{3}CV/AC^{*} \xrightarrow{\text{能量转化}} NP \xrightarrow{\text{直接光催化}}$$
 方物

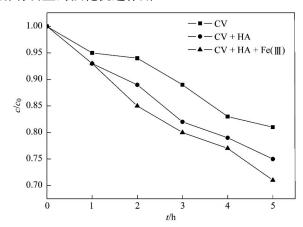
(3)光照会促使部分藻类细胞壁破碎,破碎细胞中会有脂类物质释放到水中,这些脂类物质也可能引发壬基酚的光降解:

$$NP \xrightarrow{lipid}$$
产物

- 2.2 藻/腐殖酸/铁离子引发千基酚的光降解
- **2.2.1** 一元、二元、三元体系中壬基酚的光降解对照

由于藻、腐殖酸和铁离子都为天然水环境中普遍存在的物质成分,为考察藻与其它非生物物质协

同作用的影响,本实验比较了只含藻(一元)、含藻和腐殖酸(二元)以及含藻、腐殖酸和铁离子(三元)的体系中壬基酚的光降解情况,分别配制了不同组分的反应液进行光照实验. 当pH值为6.5时,在含小球藻的一元、二元、三元体系中,壬基酚的光降解对照结果如图3所示. 实验结果显示,在含藻、腐殖酸和铁离子的三元体系中壬基酚的降解效果最明显,腐殖酸和铁离子的加入对壬基酚的光降解有明显的强化促进作用.



[NP] $_0$  = 6 mg·L $^{-1}$ , [CV] $_0$  = 2 × 10 $^9$  cells·L $^{-1}$ , [HA] $_0$  = 4 mg·L $^{-1}$ , [Fe] $_0$  = 20  $\mu$ mol·L $^{-1}$ , pH = 6.5 ± 0.1 图 3 一元、二元、三元体系中 NP 降解对照

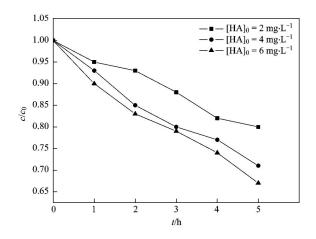
Fig. 3 Comparison of NP degradation in water with algae or algae/humic acid or algae/humic acid/ferric ions

#### 2.2.2 腐殖酸浓度变化对壬基酚光降解的影响

为考察腐殖酸对壬基酚降解的影响程度,本实验研究了腐殖酸浓度变化对壬基酚的降解的影响.在含小球藻、腐殖酸和铁离子的水溶液中,当小球藻浓度为  $2 \times 10^9$  cells·L<sup>-1</sup>,壬基酚初始浓度为 6 mg·L<sup>-1</sup>,腐殖酸浓度在  $2 \sim 6$  mg·L<sup>-1</sup>范围变化时,壬基酚的光降解率如图 4 所示.随着腐殖酸的浓度增加,壬基酚的光降解率上升.在腐殖酸浓度为 6 mg·L<sup>-1</sup>时,NP的降解率可达 33%.本实验中也同时研究了三元体系中,当藻为鱼腥藻且其浓度为  $2 \times 10^9$  cells·L<sup>-1</sup>时 NP 的光降解,研究结果表明,三元体系中,鱼腥藻对 NP 的降解促进与小球藻类似.

#### 2.2.3 铁离子对壬基酚光降解的影响

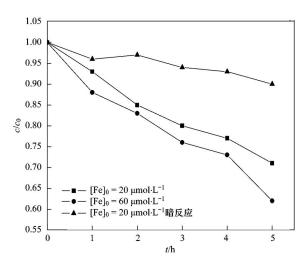
为考察铁离子对壬基酚降解的影响程度,本实验研究了铁离子浓度变化对壬基酚的降解的影响. 含小球藻、腐殖酸和铁离子的水溶液中,当小球藻浓度为  $2 \times 10^9$  cells·L<sup>-1</sup>, NP 初始浓度为 6 mg·L<sup>-1</sup>,腐殖酸浓度为 4 mg·L<sup>-1</sup>时,铁离子浓度变化对壬基酚的光降解的影响如图 5 所示. 三元体系



$$\begin{split} \left[ \text{ NP} \right]_0 = & 6 \text{ mg} \cdot \text{L}^{-1} \text{ , } \left[ \text{ CV} \right]_0 = & 2 \times 10^9 \, \text{cells} \cdot \text{L}^{-1} \text{ ,} \\ \left[ \text{ Fe} \right]_0 = & 20 \text{ } \mu \text{mol} \cdot \text{L}^{-1} \text{ , } \text{ pH} = & 6.5 \pm 0.1 \end{split}$$

图 4 腐殖酸的浓度变化对壬基酚的光降解影响

Fig. 4 Effect of the change in humic acid concentration on the photodegradation of NP



 $[NP]_0 = 6 \text{ mg} \cdot L^{-1}, [CV]_0 = 2 \times 10^9 \text{ cells} \cdot L^{-1},$  $[HA]_0 = 4 \text{ mg} \cdot L^{-1}, pH = 6.5 \pm 0.1$ 

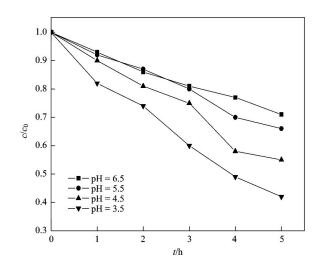
图 5 铁离子的浓度变化对壬基酚的光降解影响

Fig. 5 Effect of the change in ferric ions concentration on the photodegradation of NP

中,铁离子浓度增加,壬基酚的降解率升高.

#### 2.2.4 水溶液 pH 值变化对壬基酚光降解的影响

由于水环境中 pH 值对物质的转化有一定的影响,本实验研究了 pH 值变化对壬基酚降解效果的影响. 在含小球藻、腐殖酸和铁离子的水溶液中,当小球藻浓度为 2×10° cells·L⁻¹, NP 初始浓度为 6 mg·L⁻¹, 腐殖酸浓度为 4 mg·L⁻¹时,铁离子浓度为 20 μmol·L⁻¹时,pH 值变化对 NP 的光降解的影响如图 6 所示. 三元体系中,pH 值变化对壬基酚的光降解的影响很大,酸性条件有利于壬基酚的光降解,因为在酸性条件时,铁离子的增强促进作用



[NP] $_0 = 6 \text{ mg} \cdot \text{L}^{-1}$ , [CV] $_0 = 2 \times 10^9 \text{ cells} \cdot \text{L}^{-1}$ , [HA] $_0 = 4 \text{ mg} \cdot \text{L}^{-1}$ , [Fe] $_0 = 20 \text{ } \mu \text{mol} \cdot \text{L}^{-1}$  图 6 pH 变化对壬基酚的光降解影响

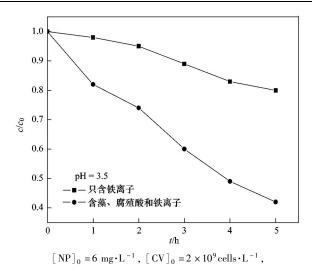
Fig. 6 Effect of pH on the photodegradation of NP

明显.

为进一步考察较低 pH 值时铁离子的影响程度,分别配制了只含铁离子和底物的反应液以及含藻、腐殖酸、铁离子和底物的反应液中壬基酚的降解情况,结果如图 7 所示,在 pH 值为 3.5 时,光照条件下,只含铁离子的水溶液中,壬基酚的光降解也很大,4 h 光照后壬基酚的光解率为 20%. 在含藻、腐殖酸和铁离子的水溶液中,4 h 光照后壬基酚的降解率为 58%,这说明酸性条件下,藻、腐殖酸与铁离子等能形成配合物并产生更多的活性氧,从而促进壬基酚的光降解.

#### 3 三元体系促进有机物降解的机制分析

本实验结果表明,含藻水溶液中加入腐殖酸、铁离子,光照后能极大促进壬基酚的降解. 这表明,藻、腐殖酸和铁离子光照后能生成某些活性物质,并且藻、腐殖酸、铁离子能形成复合物,复合物在光照后能产生活性物质,从而促进壬基酚的降解. 研究表明,腐殖质中所含有的 O、N、S 原子的基团具有能提供孤对电子的能力,因此能够与金属元素离子形成配合物、螯合物等. Piana 等[19]采用荧光发射光谱法和同步扫描光谱法研究了金属离子与腐殖质的相互作用与表征,这些研究结果表明,金属离子与腐殖质可通过相互作用形成二元复合物. 腐殖质微粒吸附污染物的机制是: 螯合、络合和离子交换吸附等. 复合作用可抑制金属以碳酸盐、磁化物、氢氧化物形式的沉淀产生.



 [HA]<sub>0</sub> = 4 mg·L<sup>-1</sup>, [Fe]<sub>0</sub> = 20 μmol·L<sup>-1</sup>, pH = 3.5 ± 0.1
 图 7 NP 在只含铁离子的水溶液中和在含小球藻/腐殖酸/ 铁离子的水溶液中的光降解对照

Fig. 7 Comparison of NP degradation in water with only ferric ions or in water with algae/humic acid/ferric ions

藻类与金属离子间也存在相互作用,其中的主 要作用是络合与离子交换,特别是多糖中的藻酸盐 与硫酸盐多糖具有显著的离子交换能力[20]. Crist 等[21,22]认为藻表面的羧基、硫酸盐组等是金属离 子交换的位点,与溶液的 pH 值有关. 藻表面包含大 量多糖和蛋白质,这其中就含有阴离子的羧基、硫 酸盐和磷酸盐等[23],这些成分可以生物吸附水溶液 中的金属离子. 藻也能以某种形式与腐殖酸结合, Pempkowiak 等[24] 研究表明, 从海水中分离出来的腐 殖物质和藻、镉同时存在的时候,腐殖物质可吸附 到藻的表面,从而降低了藻对镉的吸附. Gueguen 等[25]研究表明,当藻、腐殖酸同时存在时,金属离 子更易与腐殖酸形成配合物. 因此,溶液中金属离 子能与藻或腐殖酸形成配合物,在光照条件下能产 生更多的活性氧,从而促进有机物的降解,活性物质 的促进机制分析如下,

#### 3.1 各成分各自产生氢氧自由基的作用

藻液光照后自身能产生氢氧自由基.含可溶性有机质的水溶液在光的作用下可生成水合电子<sup>[26]</sup>,生成的水合电子能迅速地与氧反应生成 O<sub>2</sub>-.O<sub>2</sub>-可以自发或通过超氧化物歧化酶(SOD)很快歧化为 H<sub>2</sub>O<sub>2</sub>.在铁离子存在条件下,其可通过氧化还原反应产生氢氧自由基. Vaughan 等<sup>[27]</sup>研究证实天然水在有氧或无氧条件下都能产生氢氧自由基,认为氢氧自由基主要来自于腐殖物质. 因此在藻/腐殖酸/铁离子共同存在时,体系的自由基产生途径和产生量增加,因此对酚类物质的光降解促进作用增强.

#### 3.2 各成分的复合物产生氢氧自由基

Zuo 等<sup>[28]</sup>研究认为 Fe(Ⅲ)-草酸盐配合物是大气水相中的普通成分,它们吸收 UV-可见区的光能后光解生成H<sub>2</sub>O<sub>2</sub>,进而生成·OH. 因此可以推测,在含藻/腐殖酸/铁离子的水溶液中,藻、铁离子和腐殖酸等可通过相互作用形成复合物,在光照条件下通过一系列转化生成氢氧自由基<sup>[14]</sup>,氢氧自由基的生成途径增多,从而对酚类物质的光降解促进作用增强,其途径分析如下.

①铁离子和腐殖酸:

HA + Fe( 
$$\blacksquare$$
 ) → HA-Fe(  $\blacksquare$  )  $\xrightarrow{\text{light}}$  Fe(  $\blacksquare$  ) + HA• HA +  $\not$   $\longrightarrow$  H<sub>2</sub>O<sub>2</sub>

──产物(products)

HO・+ 酚类物 → 有机自由基(organic radicals) → 产物(products)

②藻/腐殖酸/铁离子:

HA + Fe( 
$$\blacksquare$$
 ) → HA-Fe(  $\blacksquare$  )  $\xrightarrow{\text{light}}$  Fe(  $\blacksquare$  ) + HA•  $\stackrel{\text{H}_2\text{O}}{\rightleftharpoons}$  H<sub>2</sub>O<sub>2</sub>

Fe (Ⅱ) + H<sub>2</sub>O<sub>2</sub> → Fe(Ⅲ) + HO· + OH<sup>-</sup> HO· + 酚类物 → 有机自由基(organic radicals) → 产物(products)

#### 4 结论

- (1)研究表明,在金属卤化物灯的光照下,含藻(鱼腥藻 A. cylindrica 和小球藻 C. vulgaris)水溶液光照后能促进壬基酚的光降解,并且随着水溶液中藻浓度增大,壬基酚的光降解效率也增大.本实验对比了含藻水溶液中壬基酚的光反应和暗反应,结果表明,光照对于壬基酚的降解起主要诱导作用.
- (2)研究了水中常见成分藻、铁离子及腐殖酸共同组成的三元水体系中壬基酚的光降解,对比研究了含藻的一元与二元、三元体系中烷基酚类物质的降解,结果显示,含藻、腐殖酸和铁离子的三元体系对酚类物质的降解促进效果更好.含藻、腐殖酸和铁离子的三元体系中,当腐殖酸浓度上升时,壬基酚的光降解增强,铁离子与腐殖酸、藻能协同促进壬基酚的光降解.在中性条件下,铁离子通过与藻(或腐殖酸)形成配合物在光照下能促进壬基酚底物的光降解;在酸性条件下,铁离子对壬基酚的光降解促进作用更明显,这是因为在pH在2~5范围

内,Fe(Ⅲ)-OH 复合物在光催化作用下会产生氢氧自由基,从而促进有机物降解转化.

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## **HUANJING KEXUE**

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